

EPA Handbook for Optical Remote Sensing

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Planning and Standards**



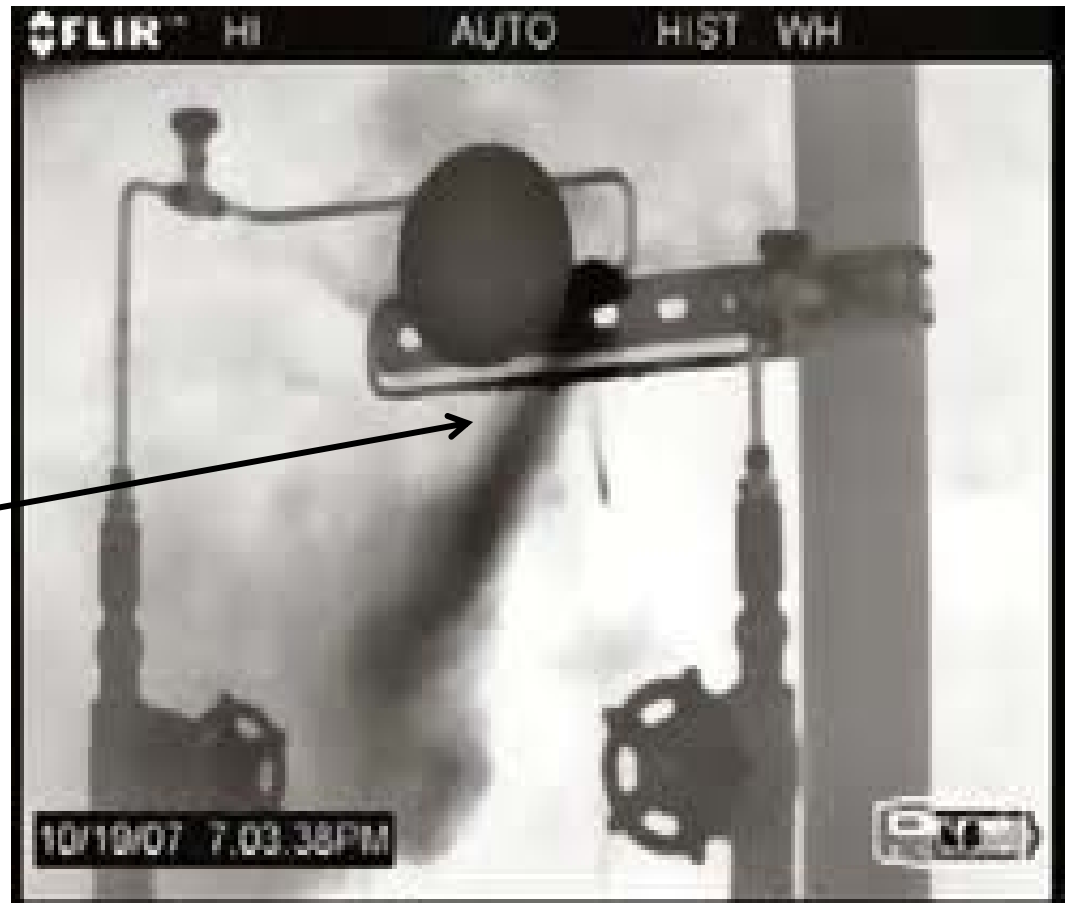
Outline

- ▶ Strategy
- ▶ What to Expect
- ▶ Chapter Description
- ▶ Deadlines and Next Steps

What do we mean by Remote Sensing?



What do we mean by Remote Sensing?



Leak Detection

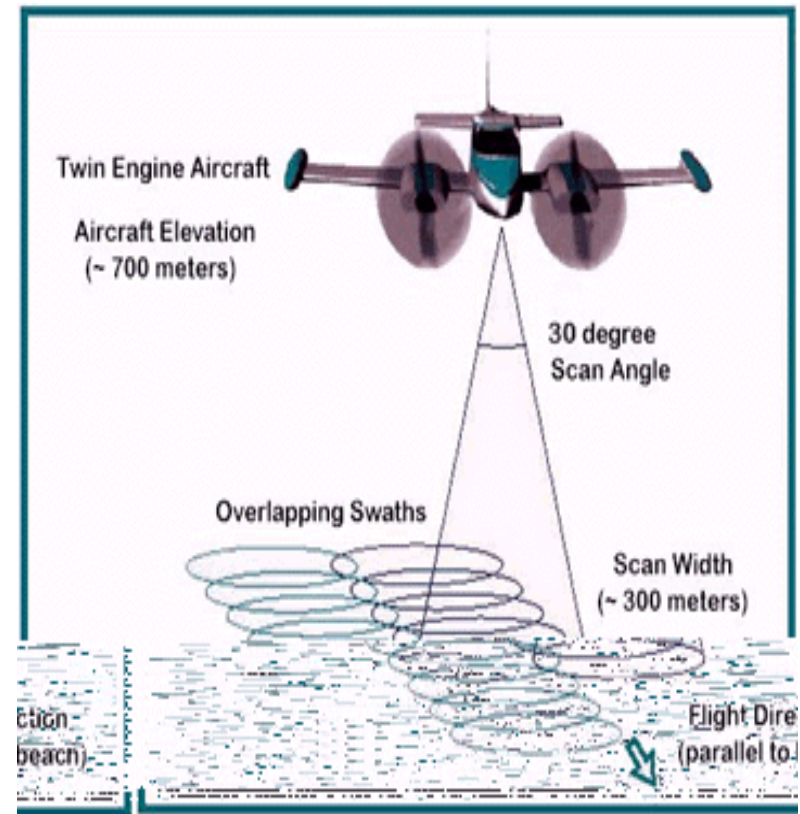
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What do we mean by Remote Sensing ?



Roadside Monitoring

Air borne LIDAR – Leak Detection



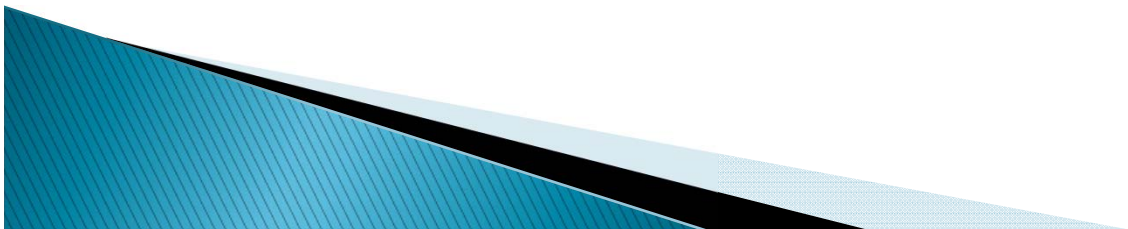
Strategy

Industry, researchers and State/Local/Federal agencies have been requesting EPA guidance

Reasoning: create a document that can be of significant use to the non-spectroscopist.

Target Audience: Engineers, scientists, monitoring professionals

Create a comprehensive, but *living* document!!



What to Expect

We will attempt to address issues related to Optical Remote Sensing

- ▶ Technology and Technique Crosswalk
- ▶ Strengths and Limitations
- ▶ Instrument Descriptions
- ▶ Lists of Target Compounds
- ▶ Method Descriptions
- ▶ Data Validation

Chapter 1 – Introduction

- ▶ Purpose
- ▶ Evolution and Updates
- ▶ EPA Quality System Summary
 - Range, Sensitivity, Measurement Errors and Uncertainty

Selecting the Right Tool for Program Objectives



Chapter 1: Technology/Technique Crosswalk

| Application | Technology |
|-------------------------------------|------------------------|
| Emission Flux Applications | |
| DIAL | LIDAR |
| SOF | |
| Emission Factor | ORS FTIR, UV-DOAS, TDL |
| bLS | ORS FTIR, UV-DOAS, TDL |
| Emission Concentration Applications | |
| ID VRPM | ORS FTIR, UV-DOAS, TDL |
| Tracer Correlation | CRDS/TDL, Mobile FTIR |
| Leak Detection Applications | |
| Optical Imaging | FTIR Camera |

Chapter 1: Strengths and Limitations

| Strengths | Limitations |
|--|---|
| Fourier Transform Infrared Spectroscopy – 2.1 | |
| Relatively low instrument cost (about \$80,000 - \$125,000) | Gas-phase water spectral interference as well as CO and CO ₂ interference ^{5,12,13} |
| FTIR equipment is fairly rugged and easily portable | Not applicable to homonuclear diatomic gases such as chlorine, oxygen, and nitrogen ^{1,2,3} |
| There are a large number of compounds that are infrared active (absorb IR light) | Weak IR absorption features for many inorganic molecules such as sulfur dioxide and nitrogen oxides ⁶ |
| Large number of compounds can be analyzed simultaneously | Infrared beam has a limited range and may not be sensitive enough to meet ambient data quality objectives. Maximum path length is on the order of 400–500 meters. |
| No gas calibration standards necessary (uses standard reference spectral library) | Multiple vertical or horizontal path measurements necessary to calculate plume flux, can require significant time and cost to set up and implement |
| FTIR can be used to locate discrete emissions hotspots at a facility/area source | Typical infrared detectors require cryogenic cooling to operate. Liquid nitrogen used for detector cooling must be refilled and maintained regularly (weekly). |
| Multi-compound coverage makes FTIR ideal for leak detection or source location where the facility being monitored has multiple compounds present (e.g., chemical plants) | Typical set-up time usually requires about 5 to 8 hours and a minimum of two people |
| Equipment can be allowed to run unattended for months at a time with remote access to check instrument operation and recover data | Single beam open-path method measures concentration along a path. The path must capture most if not all of an analyte plume to provide accurate measure of emissions. |
| No sample collection, handling, or preparation is necessary | Field implementation and data collection requires highly experienced personnel |

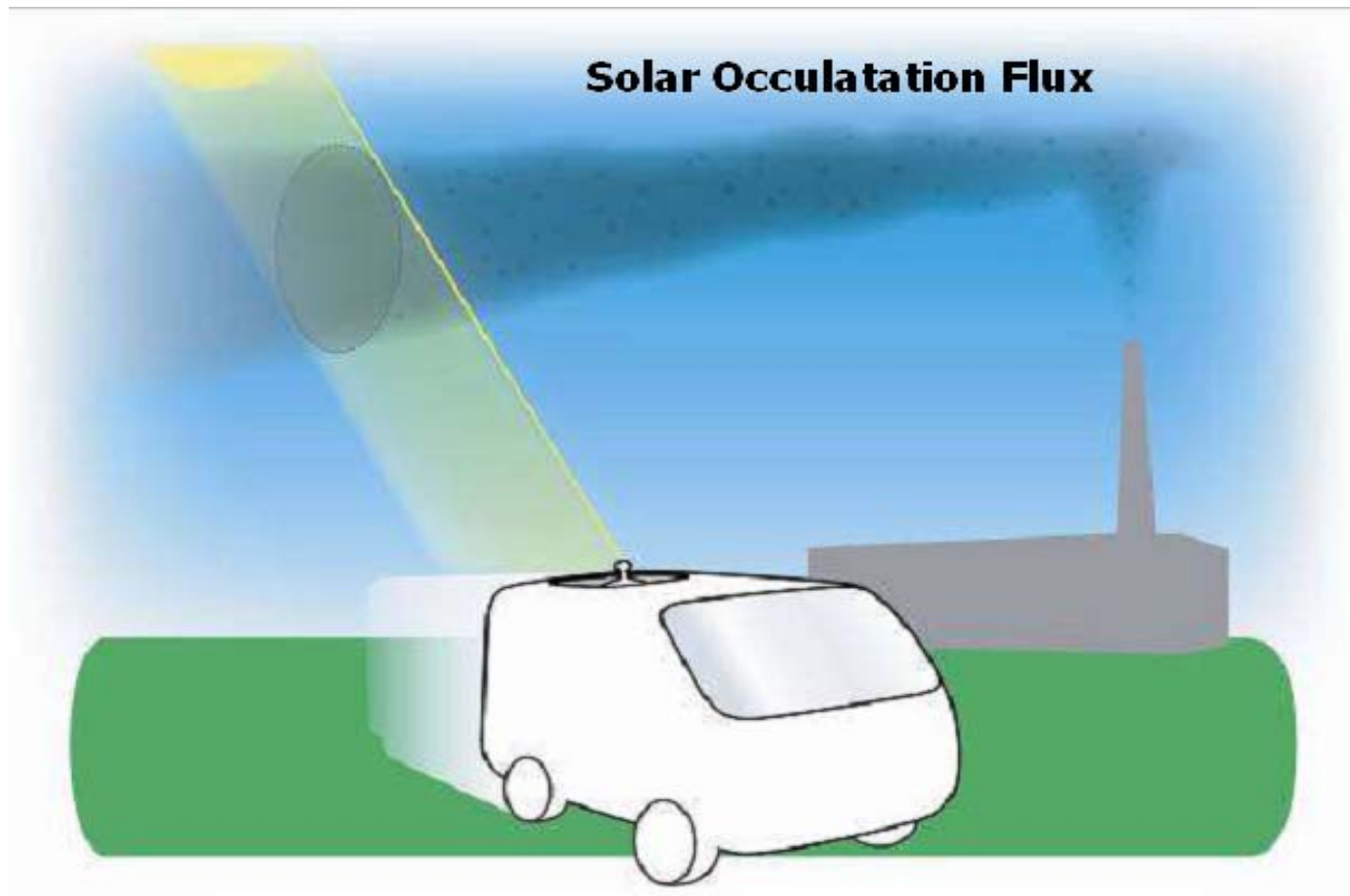
Chapter 2: Target Compounds

| Laser Type | λ Range (nm) | Target Compounds |
|--------------------------|----------------------|--|
| AlGaInP | 630-690 | NO ₂ |
| AlGaAs | 750-1000 | O ₂ , NH ₃ |
| Vertical Cavity | 650-1680 | H ₂ O, C ₂ H ₂ , HF, H ₂ S, O ₂ , H ₂ O, NH ₃ |
| Antimonide* | 2000-4000 | CO, CO ₂ , NO, N ₂ O, CH ₄ , HCl, HBr, H ₂ CO |
| Quantum Cascade** | 4000-12000 | H ₂ O, CO, CO ₂ , NO, NO ₂ , N ₂ O, SO ₂ , C ₂ H ₂ , HCN, NH ₃ , PH ₃ , O ₃ |
| Lead-salt** | 3000-30000 | H ₂ O, CO, CO ₂ , NO, NO ₂ , N ₂ O, SO ₂ , CH ₄ , C ₂ H ₂ , HCl, HBr, HCN, NH ₃ , H ₂ CO, PH ₃ , O ₃ |

Chapter 2 - Compounds and Detection

| Species (Tuneable Diode Lasers) | Approximate near-IR λ (nm) | Reported Detection Limit (ppm-m) |
|---------------------------------|------------------------------------|----------------------------------|
| ammonia | 760, 1500 | 0.5-5.0 |
| carbon monoxide | 1570 | 40-1,000 |
| carbon dioxide | 1570 | 40-1,000 |
| hydrogen chloride | 1790 | 0.15-1 |
| hydrogen cyanide | 1540 | 1.0 |
| hydrogen fluoride | 1310 | 0.1-0.2 |
| hydrogen sulfide | 1570 | 20 |
| methane | 1650 | 0.5-1 |
| nitric oxide | 1800 | 30 |
| nitrogen dioxide | 680 | 0.2 |
| oxygen | 760 | 50 |

Chapter 3 – Measurement Approaches



Chapter 4 – Ancillary Data

- ▶ Types of data that is typically collected

- ▶ Meteorological Measurements

 - How is Met data used to calculate flux
 - Monitoring location siting

- ▶ Process information: Emission factors, through-put, stack height, exit velocities

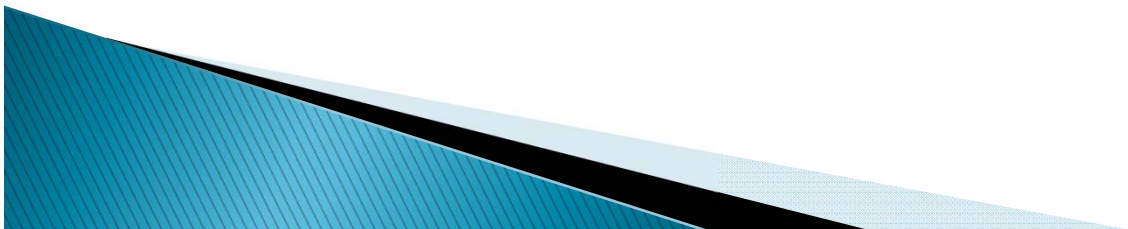
- ▶ Other useful data: GPS



Timeline

- ▶ OAQPS – requested 2010 fiscal year funds
- ▶ Work Assignment just ended
- ▶ Sections 1 – 5 are written
- ▶ All sections are under external and internal review

First Draft must be finished by December 2010



Timeline

- ▶ Due to time constraints:
- ▶ Modified the original Table of Contents
- ▶ Focusing on the “flux” determination methods
- ▶ Intend to expand document to include:
 - Ambient – Near roadway monitoring
 - Mobile Monitoring
 - Fence–line Monitoring



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